

Kurze Mitteilungen

Maximalumfang: 6 Schreibmaschinenseiten (alles inbegriffen). Bis zum 5. des Monats bei der Redaktion eingehende Manuskripte können günstigenfalls am 15. des folgenden Monats veröffentlicht werden.

Some Reactions on 2-Chloroacetylphenothiazine *

H. H. Zoorob **, W. S. Hamama

Faculty of Science, El-Mansoura University, Chemistry Department, Egypt

M. T. El-Wassimi, M. M. Abbasi

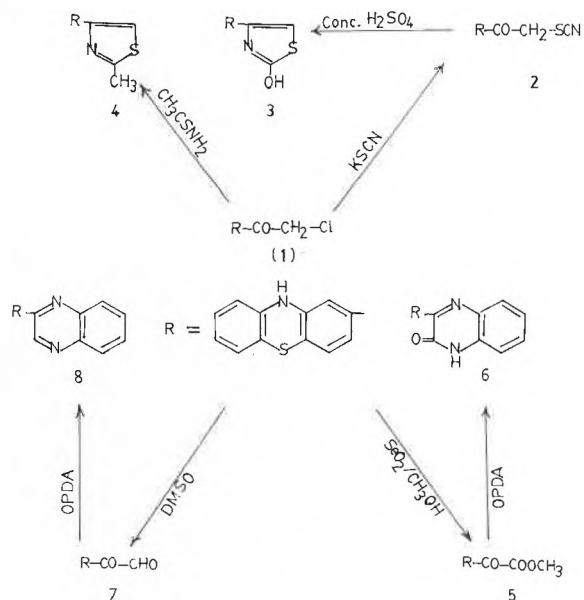
Faculty of Science, Assuit University and Faculty of Science, Tanta University, Chemistry Department, Egypt.

Abstract:

The possibility of using 2-chloroacetylphenothiazine (1) for the synthesis of thiazole moieties has been studied. In addition, we have investigated the conversion of 1 to methyl phenothiazine-2-glyoxylate (5) and phenothiazine-2-glyoxaldehyde (7) by the action of $\text{SeO}_2/\text{CH}_3\text{OH}$ and dimethylsulphoxide respectively.

The interesting pharmacological properties of phenothiazine prompted us to introduce new substituents into this molecule. One of these variations is the introduction of heterocyclic moieties in the 2-position of the phenothiazine [1]. Therefore, we studied the possibility of using 2-chloroacetylphenothiazine (1) to synthesize a thiazole system containing in the 4-position a phenothiazin-2-yl group.

Thus, 2-chloroacetylphenothiazine (1), prepared according to method of *Burger & Clements* [2], readily condenses with potassium thiocyanate in ethanol to give 2-(thiocyanato-acetyl)-phenothiazine (2). Treat-



* Received August 19, 1980.

** Author to whom correspondence may be addressed.

Scheme 1

ment of 2 with concentrated sulphuric acid gave 2-(2-hydroxy-4-thiazolyl)-phenothiazine (3). The IR spectrum of 3 showed absorption bands at 3100 (OH), 3310 (NH) and at 1590 cm^{-1} (thiazole moiety).

Compound 1 reacts readily with thioacetamide in boiling alcohol to give 2-(2-methyl-4-thiazolyl)-phenothiazine (4). It may be mentioned its isomer, 2-(4-methyl-2-thiazolyl)-phenothiazine, was synthesized by Rhône-Poulenc [3]. Structure 4 was established through its analysis and IR spectrum, which showed peaks at 1590 (thiazole moiety), 3360 (NH) and at 1480, 1360 cm^{-1} (CH_3).

Methylphenothiazine-2-glyoxylate (5) was synthesized by oxidation of 2-chloroacetylphenothiazine (1) with selenium dioxide in absolute methanol according to the synthesis of ethyl phenylglyoxylate [4]. Compound 5 was not obtained in crystalline form and was therefore characterized as its quinoxalinone derivative 6, the 2-(2-oxo-1,2-dihydro-3-quinoxaliny)-phenothiazine (6), made by condensation of o-phenylenediamine (OPDA) with 5. The IR spectrum of the quinoxalinone derivative 6 showed absorptions at 1615 ($\text{C}=\text{O}$ amide) and at 3360 cm^{-1} (NH), any signal of the starting ester is absent. Its mass spectrum shows the molecular ion M^+ at m/e 343 which is in good agreement with the proposed structure.

The use of 2-chloroacetylphenothiazine (1) for the synthesis of phenothiazine-2-glyoxylaldehyde (7) was studied. This glyoxylaldehyde 7 was obtained in crystalline form by the action of dimethylsulphoxide on 1, according to Saikachi's procedure [5]. Treatment of 7 with o-phenylenediamine afforded 2-(2-quinoxaliny)-phenothiazine (8). The thiosemicarbazone derivative of 7 was prepared by reacting it with thiosemicarbazide. The IR spectrum of 7 shows at 1580 (aldehydic group) and at 1610 cm^{-1} ($\text{C}=\text{O}$), while in 8 these signals could not be seen.

Experimental

Melting points (uncorrected) were taken in open capillary tubes by the use of *Gallenkamp* electric melting point apparatus. Infrared spectra were performed on Pye Unicam Infracord Spectrophotometer model SP 2000 using KBr. The mass spectrum was taken on an A.E.I. MS 9 instrument.

2-(Thiocyanato-acetyl)-phenothiazine (2):

A solution of 2-chloroacetylphenothiazine (1) (2.75 g, 10 mmole) in ethanol (40 ml) was added to a solution of potassium thiocyanate (1.54 g; 14 mmole) in ethanol (5 ml). The mixture was heated for an hour with stirring on a boiling steam bath. The precipitate was isolated by filtration and recrystallized from benzene to give 2 as orange crystals, m.p. 215–216°C (quantitative yield).

Analysis: $\text{C}_{15}\text{H}_{10}\text{N}_2\text{OS}_2$ (298.39)

required: C: 60.38; H: 3.38; S: 21.49

found: C: 60.64; H: 3.48; S: 21.67.

Cyclization of 2: 2-(2-hydroxy-4-thiazolyl)-phenothiazine (3):

A mixture of 2 (2.09 g; 7 mmole) in glacial acetic acid (40 ml) and concentrated sulphuric acid (0.3 ml) was heated with stirring for 30 minutes on a steam bath and then cooled. The precipitate

was recrystallized from isopropanol-benzene mixture (1:1) to give 3 as yellow-brown crystals, m.p. 300°C (90% yield).

Analysis: $\text{C}_{15}\text{H}_{10}\text{N}_2\text{OS}_2$ (298.39)

required: C: 60.38; H: 3.38; S: 21.49

found: C: 60.72; H: 3.13; S: 20.99.

2-(2-Methyl-4-thiazolyl)-phenothiazine (4):

A solution of 2-chloroacetylphenothiazine (1) (0.83 g; 3 mmole) and thioacetamide (0.23 g; 3 mmole) in ethanol (20 ml) was heated with stirring on a steam bath for an hour, and then cooled. The precipitate formed was recrystallized from ethanol to give 4 as yellow crystals, m.p. 183°C (quantitative yield).

Analysis: $\text{C}_{16}\text{H}_{12}\text{N}_2\text{S}_2$ (296.42)

required: C: 64.83; H: 4.08; S: 21.64

found: C: 64.62; H: 4.28; S: 21.33.

2-(2-Oxo-1,2-dihydro-3-quinoxaliny)-phenothiazine (6):

A mixture of 2-chloroacetylphenothiazine (1) (13.7 g; 50 mmole), selenium dioxide (5.85 g; 50 mmole) and absolute methanol (50 ml) was stirred and refluxed for 10 hours. The selenium was removed and the filtrate was evaporated, leaving methyl phenothiazine-2-glyoxylate (5) as an oil.

A solution of o-phenylenediamine (1 g; 10 mmole) in ethanol (20 ml) was added to one tenth of the volume of the crude glyoxylate 5 and the mixture was heated on a steam bath for 30 minutes. After dilution with water the precipitate formed was crystallized from benzene to give 6 as brown-red crystals, m.p. 206°C (60% yield).

Analysis: $\text{C}_{20}\text{H}_{13}\text{N}_3\text{OS}$ (343.41)

required: C: 69.95; H: 3.82; N: 12.24

found: C: 70.25; H: 3.65; N: 12.47.

Phenothiazine-2-glyoxylaldehyde (7):

A solution of 2-chloroacetylphenothiazine (1) (0.826 g; 3 mmole) in dimethylsulphoxide (7 ml) was heated for 4 hours on a steam bath and allowed to stand overnight. After the addition of water (50 ml) the precipitate formed was recrystallized from benzene and gave red crystals, m.p. 194°C (70% yield). Its thiosemicarbazone derivative was crystallized from alcohol, yellow crystals, m.p. 182°C.

Analysis: $\text{C}_{14}\text{H}_9\text{NO}_2\text{S}$ (255.30)

required: C: 65.87; H: 3.55

found: C: 65.49; H: 3.73

2-(2-Quinoxaliny)phenothiazine (8):

A solution of o-phenylenediamine (1.08 g; 0.01 mole) in ethanol (20 ml) was added to glyoxylaldehyde derivative 7 (2.55 g; 0.01 mole) in ethanol (5 ml). The mixture was heated on a steam bath for 30 minutes and then diluted with water. The crystalline mass separating was filtered off and recrystallized from ethanol to give 8 as yellow crystals, m.p. 230°C (77% yield).

Analysis: $\text{C}_{20}\text{H}_{13}\text{N}_3\text{S}$ (327.41)

required: C: 73.37; H: 4.00; N: 12.84

found: C: 73.66; H: 4.32; N: 12.59.

References

- 1 A. R. Katritzky and A. J. Boulton: "Advances in Heterocyclic Chemistry", vol. 9, Academic Press, New York, (1968) p. 321.
- 2 A. Burger and J. B. Clements: *J. Org. Chem.*, 19 (1954) 1113.
- 3 Rhône-Poulenc, Belg. Pat. 612886 (1962); C.A., 59 (1963) 1653c.
- 4 J. P. Schaefer and E. J. Corey: *J. Org. Chem.*, 24 (1959) 1827.
- 5 H. Saikachi and J. Matsuo: *Yakugaku Zasshi*, 88 (1968) 1306.