

$C_{12}H_{16}NO_3F_3$; C 51.61, H 5.78, N 5.02, O 17.19, F 20.41; found: C 51.69, H 5.92, N 5.24, F 20.38.

(R)-1-Phenylethylammonium (*S*)-4,4,4-Trifluoro-3-hydroxybutanoate ((S,R)-2): $[\alpha]_D = -2.3$ ($c = 1.09$, EtOH).

(R)-4,4,4-Trifluoro-3-hydroxybutanoic Acid ((+)-1): M.p. 43.8–44.0°. $[\alpha]_D = +15.1$ ($c = 7.27$, EtOH). IR (KBr): 3460, 3180, 1735, 1275, 1180, 1130. 1H -NMR (CD_3OD , 300 MHz): 4.40 (*ddq*, $J_1 = 3.3$, $J_2 = 9.7$, $J(H,F) = 7.1$, CH(CF_3)); 2.67 (*dd*, $J_1 = 16.0$, $J_2 = 3.3$, CH_AH_B); 2.50 (*dd*, $J_1 = 16.0$, $J_2 = 9.7$, CH_AH_B). ^{13}C -NMR (CD_3OD , 75 MHz): 173.1 (*s*); 126.7 (*g*, $J(C,F) = 281$, CF_3); 68.2 (*g*, $J(C,F) = 32$, CH); 36.6 (CH₂). ^{19}F -NMR (CD_3OD , 282 MHz, Ref. $CFCl_3$): -79.9 (*d*, $J(H,F) = 7.2$). MS: 159 (54), 141 (84), 120 (49), 89 (86), 71 (95), 69 (42), 43 (100). Anal. calc. for $C_4H_5O_3F_3$: C 30.39, H 3.19, O 30.36, F 36.05; found: C 30.16, H 3.23, F 36.0.

(S)-4,4,4-Trifluoro-3-hydroxybutanoic Acid ((-)-1): $[\alpha]_D = -15.0$ ($c = 6.58$, EtOH).

Methyl (R)-4,4,4-Trifluoro-3-hydroxybutanoate: $[\alpha]_D = +21.0$ ($c = 4.78$, $CHCl_3$). IR (film): 3460, 2960, 1730, 1440, 1275, 1170, 1130. 1H -NMR ($CDCl_3$, 300 MHz): 4.51–4.40 (*m*, CH(CF_3)); 3.77 (*s*, CH_3O); 2.75 (*dd*, $J_1 = 16.8$, $J_2 = 4.2$, CH_AH_B); 2.68 (*dd*, $J_1 = 16.8$, $J_2 = 8.2$, CH_AH_B). ^{13}C -NMR ($CDCl_3$, 75 MHz): 171.3 (*s*); 124.5 (*g*, $J(C,F) = 281$, CF_3); 67.3, (*g*, $J(C,F) = 32$, CH); 52.5 (CH₃); 34.7 (CH₂). ^{19}F -NMR ($CDCl_3$, 282

MHz, Ref. $CFCl_3$): -80.4 (*d*, $J(H,F) = 7.1$). MS: 173 (0.5), 141 (48), 103 (40), 69 (26), 59 (40), 43 (100), 28 (78), 15 (66). Anal. calc. for $C_5H_7O_3F_3$: C 34.90, H 4.10, O 27.89, F 33.11; found: C 34.77, H 4.27, F 32.9. Methyl (*S*)-4,4,4-Trifluoro-3-hydroxybutanoate: $[\alpha]_D = -20.9$ ($c = 2.11$, $CHCl_3$).

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- [1] The attention paid to this class of compounds is illustrated by the number of contributions about this topic in the *International Chemical Congress of Pacific Basin Societies* (12% of the presentations in the Organic Division!): *Pacifichem '89*, Honolulu, Hawaii, 17.–22.12.1989.
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[6] Salt molecules or the dissociated components in the gas phase? Cf. C. G. de Kruif, *J. Chem. Phys.* **1982**, 77, 6247.

[7] At 0.001 Torr, the CH_3 , CCl_3 , and CF_3 derivatives start subliming at 40, 40, and 20°, respectively.

[8] Japanese Patent Application Jpn. Kokai Tokkyo Koho JP 6403,154, S. Mihashi, H. Kondo (CA: 1989, 111, 31747 p).

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[10] Determined by esterification with CH_2N_2 and ^{19}F -NMR spectroscopy of the Mosher ester [4].

Groupe d'Analyse Chimique Interdisciplinaire de l'EPFL, GACHI*: 10e Journée d'Analyse: ICP-MS

Raymond Houriet**

The 10th meeting was held on November 28, 1989, sponsored by Perkin-Elmer and VG Instruments. Its topic was the coupling between emission spectroscopy and mass spectrometry: ICP-MS. The state of the art in the method was introduced by Prof. Mermet. Recent developments were described by the manufacturers and applications were presented by users of the method, including contributions of three (out of the four) Swiss laboratories equipped with the method. Three articles discuss the limits of detection in emission vs. ICP-MS (E. Poussel and J. M. Mermet), the applications of ICP-MS in an industrial analytical center (H. Baumann)*** and the impact of ICP-MS on studies in groundwater typology (A. Parriaux and J. D. Dubois)***.

* For further information, see *Chimia* 1988, 42, 398.

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