pecially the smaller oligolides can also be

obtained directly from the polymer: under

carefully controlled acid-catalyzed conditions, the triolid is prepared from PHB in ca. 50% yield [14][19]. A number of X-ray crystal structures [14][18] reveal the following facts: i) the larger oligolides (≥ 6 HB units) occur in several forms

which differ in the folding of the rings (conformation) in the crystals; *ii*) the fold-

ed rings contain structural units from which known [20] and hitherto not detected hel-

ical structures of the linear polymer can be modelled [14][21]. A data bank of bond

lenghts, bond angles, and dihedral angles for further modellings of polyester structures has become available. The crystal structures of the oligolides and their solidstate NMR spectra have been compared

[22]. Finally, enzymatic degradation

Chimia 47 (1993) 102-103
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Poly[(R)-3-hydroxyalkanoates] – The Fourth Family of Biopolymers: Contributions of an Organic Chemist

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Originally, we considered PHB as a source of (R)-3-hydroxybutanoic acid (HB), a chiral synthetic building block [1], the enantiomer of the yeast reduction product of 3-oxo-butanoate [2]. We learned how to modify the structure of this simple molecule at each and every position of the C_4 skeleton [3–5] and to apply the principles developed with it to other 3-hydroxy-[6][7] and also to 3-amino-carboxylic acids [8][9].

Over the years, we became more and more interested in biological, biochemical, medicinal, and material-science aspects of PHB and the other poly(hydroxyalkanoates) (PHAs). PHB occurs in nature not only as the high-molecularweight microbiological storage material (s-PHB, m.w. ca. 106 Dalton) but also as an oligomer of 100-150 units (c-PHB, m.w. around 104 Dalton) supposed to form complexes with various other materials. In her pioneering work, Reusch has shown that c-PHB is present in cell walls of prokaryotic and eukaryotic organisms [10][11], as well as in human blood serum [12]. The proposed function of c-PHB in biological systems called for fundamental synthetic and structural investigations!

We have developed a simple exponential fragment coupling method for preparing linear oligomers of very narrow molecular-weight distribution [13][14] which gave, for instance, a 93-mer 1 with $M_w/M_n \le 1.001$ [15], and we found new methods

$$n = 3-10$$

$$n = 3-10$$

of degrading PHB to oligomer fractions indicative of the dimension of folding in PHB crystallites [16][17]. Laser-desorption mass spectroscopy and gel-permeation chromatography were used for molecular-weight determinations, and the synthetic materials served as standards to calibrate these methods [15].

Cyclic oligomers (oligolides) 2 of 3-hydroxybutanoic acid were prepared from the monomer, using macrolactonization methods. Depending on the conditions, different ring sizes were preferred, and pure compounds containing three to ten HB units were isolated [13][14][18]. Es-

(*Pseudomonas delafieldii*) of the cyclic oligomers has been found to occur [23] contrary to expectation (so far enzymes have only been shown to degrade PHB from the *termini* of the chain).

Complexation and transport of ions through organic phases by various linear and cyclic HB oligomers have been studied [24]. Alkali and alkaline-earth ions are transported (nonspecific lipophilic order). A crystal structure reveals a first example of an ester-crown complex between Na⁺ and the triolide, with an intriguing geometry [19].

The work presented at the Interlaken Meeting has been done by a group of enthusiastic coworkers: Andreas Brunner, H. Michael Bürger, Urs Lengweiler, Roland Meier, Hans-Martin Müller, Dietmar A. Plattner, Ingrid Schwegler (see also names in the list of references).

An extended abstract of a previous lecture on this topic has been published [25], and an extensive *review article* entitled 'Poly(hydroxyalkanoates), the Fifth Class of Physiologically Important Organic Biopolymers?' is appearing in the April 1993 issue of *Angew. Chem.* (260 references!). Reprints of the article or copies of the corresponding issue of the journal will be available at the conference.

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