methodology by using 3-lithiated 3-cyanophthalides which not only greatly increased the reaction workup time (< 1 h compared to overnight in the case of Sammes lithiumphthlalide-mediate reaction), but also because of readily availability using Swenton's elegant methodology [6], enables one to prepare a wide variety of anthraquinones, such as methyl ethers of the natural products chrysophanol [7], helmintho-sporin [7], morindpaparvin [8], and biologically active azaanthraquinones [9]. Recently, we have prepared hexacene quinones 11 by the reaction of 4bromopyrene and cyanophthalides [10] (Scheme 3).

Other dipolar nucleophiles have been discovered that react with arynes to give polycyclics. E.g., as shown in Scheme 4, the generation of another equiv. of aryne in the presence of α -lithiated rearranged nitriles 12 and 7 from the reaction of 2bromo-1,4-dimethylbenzene and 2-bromo-4-methylanisole yields the respective 9imino derivatives of 10-arylanthrones 13 [11] and 14 [12]. 10-Hydroxy-9-anthracenecarbonitriles [13] can be prepared similarly by generating arynes in the presence of α -lithiated o-(methoxycarbonyl)arylacetonitriles, with no apparent restriction on the nature of the aryne. 10-Amino-9anthracenecarbonitriles can also be prepared by the reaction of α -lithiated α cyano-o-tolunitrile with methoxy substituted arynes [14], 1-naphthalynes [15] and 1-pyrynes [12]. However, it yields rearranged dinitriles when treated with 3,6dimethoxybenzyne, 9-phenanthryne [16] and 4-pyryne [12]. The 10-amino- and 10aminoanthracenes are powerful fluores-

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cent materials, whose excitation wavelengths occur from 320 to 360 nm and emission wavelengths range from 520 to 560 nm [12].

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Novel Perylene Derivatives as Highly Photostable Fluorescent Dyes

Heinz Langhals*

Abstract. Perylene-3,4-dicarboxylic imides are prepared by a new decarboxylizing condensation with moderately sterically hindered primary amines. Perylene-3,4-dicarboxylic anhydride is prepared *via* a saponification reaction of the dicarboximide and is a starting material for a number of new types of chromophores. A second route to novel perylene derivatives is a partial hydrolysis of perylene dyes and a condensation with diamines. The dyes thus obtained are orange to red fluorescing in solution.

1. Introduction

Perylene-3,4:9,10-tetracarboxylic bisanhydride (1, C.I. No. 71127) is prepared in an analytically pure state on technical scale and it is the starting material for the preparation of perylene dyes (perylene-3,4:9,10-tetracarboxylic bisimides, 2) [1]. The perylenes 2 are well-known as highly photostable pigments and fluorescent dyes; for a review see [2]. An essential point for the lightfastness and inertness of 2 is the

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methodology by using 3-lithiated 3-cyanophthalides which not only greatly increased the reaction workup time (< 1 h compared to overnight in the case of Sammes lithiumphthlalide-mediate reaction), but also because of readily availability using Swenton's elegant methodology [6], enables one to prepare a wide variety of anthraquinones, such as methyl ethers of the natural products chrysophanol [7], helmintho-sporin [7], morindpaparvin [8], and biologically active azaanthraquinones [9]. Recently, we have prepared hexacene quinones 11 by the reaction of 4bromopyrene and cyanophthalides [10] (Scheme 3).

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linkage of two units of the strongly resonance stabilized cyclic carboxylic bisimide structure element (3) to the perylene skeleton in *peri*-position.

The anhydride 1 is predominantly used for the preparation of 2, but it is also a

valuable starting material for other derivatives with a perylene structure element. This has been reported for the preparation of carboximides with five-membered rings [3], and will be demonstrated for a number novel fluorescent dyes.

2. Results

2.1. Perylene-3,4-dicarboxylic Imides 4

Whereas 1 is a well established class of dyes there are only a few derivatives of the related 4, because there is no preparation procedure for 5. However, if moderately sterically hindered primary amines like 2,5-di(*tert*-butyl)aniline are condensed with 1 in presence of H_2O a decarboxylation proceeds with the condensation to the imide 4a in 50% yield.

The imides 4 are hydrolysed with KOH in *tert*-butyl alcohol to 5. The anhydride 5 is a general starting material for 4 and other perylene derivatives. The pure anhydride 5 is a red pigment with an intense solid-state fluorescence; the UV/VIS spectra are shown in the *Figure*.

The dyes 4 are extraordinarily photostable: in solution more even than a factor of 20 (!) compared to 2. This may be a consequence of the stabilizing structure element 3. The dyes are highly fluorescent. Some derivatives

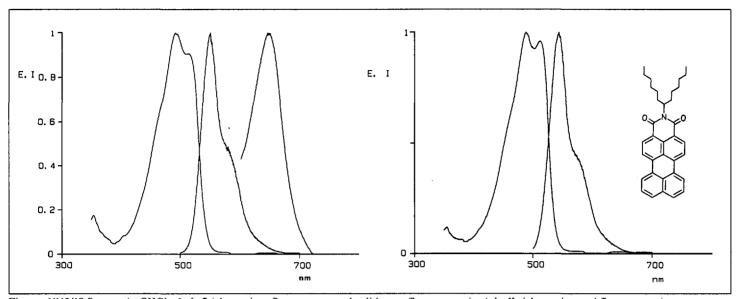


Figure. UV/VIS Spectra in CHCl₂. Left: 5 (absorption, fluorescence and solid-state fluorescence); right 4b (absorption and fluorescence).

of 4 exhibit an intense solid-state fluorescence.

2.2. Peryleneamidineimides 8 or 9

Many properties of 2 can be controlled by the substituents R, but not the UV/VIS spectra in solution [4] because of nodes in the orbitals HOMO and LUMO at the Natoms [5]. On the other hand, little is achieved by altering the basic chromophore of 2. A possibility of shifting the UV/VIS spectra is an exchange of carbonyl groups of the imide structure 3 by the related imino group. This has been done in a preceding work [6] for one carbonyl group of each imino ring. A bathochromic shift was obtained, but the fluorescence quantum yields went down to 60% or even lower.

A more moderate altering of the chromophore of 2 is the exchange of one single carbonyl group to an imino group [7]. This can be attained by a partial alkaline saponification of 2 to 7 and a condensation of 7 with diamines to amidineimides like 8 or 9. These are intensely orange to red fluorescing photostabe dyes.

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Malononitrile, a Synthetic Tool for New Chromophoric Systems

Jean-Marie Adam*, Peter Sutter, and Tammo Winkler

Abstract. New near-IR-absorbing dyes were synthesized by condensation of the dicyanovinyl derivative of 1-acetyl-3-oxo-2,3-dihydroindole with nitroso compounds. The resulting products were found to be mixtures (ca. 9:1) of the (E)- and (Z)-isomers. The more bathochromic (E)-form could be converted into the (Z)-form by heating in methanol. At higher temperature new cyclization products were obtained. A mechanism of the isomerisation and cyclization is proposed.

1. Introduction

At the present time, there is current interest in the development of near-IR absorbing dyes for use e.g. in diode-laser optical storage systems. Several investigations have been reported on the synthesis and properties of IR dyes suitable for electro-optical applications [1]. Methine dyes are widely applied in this area because their absorption bands can be easily shifted into the longwave region by the suitable choice of dye end-groups or by extending the methine chain. In this re-

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