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# Highlights of Analytical Chemistry in Switzerland

# *In situ* Element-Specific and Time-Resolved Investigation of Micro-Corrosion Processes

Nadzeya Homazava\*<sup>ab</sup>, Andrea Ulrich<sup>a</sup>, and Urs Krähenbühl<sup>b</sup>

\*Correspondence: N. Homazava<sup>a</sup>, Tel.: + 41 44 823 43 54, Fax: +41 44 823 40 41, E-mail: nadzeya.homazava@empa.ch

<sup>a</sup>EMPA, Swiss Federal Laboratories for Materials Testing and Research, Ueberlandstrasse 129, CH-8600 Dübendorf

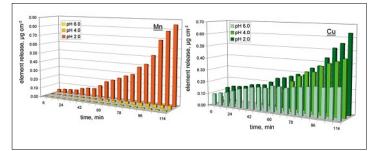
 $^{\mathrm{b}}\textsc{University}$  of Bern, Department of Chemistry and Biochemistry, Freiestrasse 3, CH-3012 Bern

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Detailed information on corrosion processes provides the key to effective prediction and minimization of corrosion damages. The initiation stage of material decomposition plays a special role, since the corrosion often starts at the weakest locations such as surface defects, grain boundaries, segregations or inclusions. How-



Corrosion is an economic issue since it destroys material goods



Time-resolved dissolution behavior of Mn and Cu in AA 6111 using 0.1 M NaCl corrosive media at different pHs

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Please contact: Dr. Veronika R. Meyer, EMPA St.Gallen, Lerchenfeldstrasse 5, 9014 St.Gallen Phone: 071 274 77 87, Fax: 071 274 77 88, Mail to: veronika.meyer@empa.ch

ever, surface analysis or electrochemical methods commonly used in corrosion research (*e.g.* electrochemical methods, SEM-EDX, *etc.*) cannot present local element-specific and online *in situ* information at the same time.

As a solution a technique for localized element-specific investigations of corrosion processes has been developed. The technique is based on an adjustable online microflow-capillary set-up especially designed for local *in situ* experiments at trace and ultratrace concentration levels. The capillary is online connected *via* flow injection (FI) analysis system to an inductively coupled plasma mass spectrometry ICP-MS. FI allows a transient sample introduction, whereas ICP-MS is designed for highly sensitive multi-element quantification.

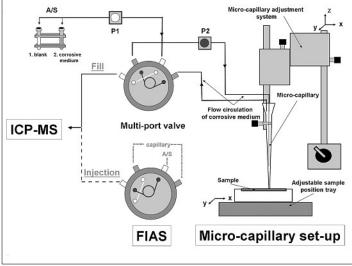
The efficiency of the developed technique could be proved by corrosion susceptibility analysis of a commercial aluminum alloy. The influence of various factors such as exposure time or pH value of corrosive media on the element-specific dissolution rates was studied in alloy AA 6111. This information is especially valuable for alloying elements present in the alloy in sub-percent quantities, which could also be detected in very low concentrations in the solution as *e.g.* Cu and Mn. The element-specific investigation of corrosion behavior of AA 6111 revealed a relatively high release of the secondary alloying element Cu in the studied pH range. **New insights into the behavior of copper during the corrosion process, not fully understood so far, can be obtained with the newly developed** *in situ* **experiments.** 

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Principle of the novel microcapillary FI-ICP-MS set-up